**Lab: Boiling Point Elevation**

**Prelab Discussion:** Describe boiling point elevation

**Purpose:**  To find the molarity of a NaCl aqueous solution (NaCl (aq)) that has a boiling point of 104˚C.

Also, create a graph showing the relationship between mass of salt added to boiling point of solution.

The independent variable is mass of salt and dependent is boiling point.

**Materials:** Water, NaCl

**Equipment:** You need to decide which equipment is needed.

**Procedure:** Design a procedure for reaching your goal.

**Obs. / Data:** record you results

You will need to record the amount of water and the amount of salt used.

**Calculations:** Show all your work for calculating molarity

**Conclusion:**  Any interesting findings.